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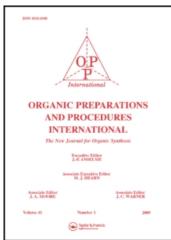
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Convenient One-Step Dialkylation of Diethyl Malonate With Sodium Hydride

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- 5. All chemicals were commercial reagent grade and were used without purification. Melting points were taken on a Thomas-Hoover apparatus in open capillaries and are corrected.

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Convenient One-Step Dialkylation of Diethyl

Malonate With Sodium Hydride.

Submitted by: Edwin M. Kaiser, Janet A. Fries and Walter J. Simonsen, Jr. Department of Chemistry University of Missouri Columbia, Missouri 65201

The authors describe the title procedure using sodium hydride in tetrahydrofuran (THF). Thus, the di-n-butyl (86%), di-n-octyl (66%), di-benzyl (98%) and di-2-phenethyl (49%) diethyl malonates were obtained in the given yields from the respective halides. Cyclizations using 1,4-dibromobutane and 1,3-dibromopropane afforded the cyclopentyl (I 63%) and

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I III cyclobuty1 (II 45%) derivatives. Significantly, the procedure may be applicable to related compounds; for example, di-n-butylmalononitrile III was similarly obtained from malononitrile (76%).

EXPERIMENTAL

General Procedure. To a refluxing slurry of four equivalents of sodium hydride² in 200 ml of anhydrous THF was added dropwise, during one hour, a solution of 0.1 m of diethyl malonate and 0.4 m of the alkyl halide (or 0.2 m of the alkyl dihalide) in 100 ml of THF. The mixture was refluxed for three more hours, cooled to 0° and hydrolyzed by cautious dropwise addition of 100 ml of water. Work up of the organic layers followed by vacuum distillation produced the desired compounds which exhibited correct boiling points^{1a} and consistent nmr spectra.

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